Qualification Accredited



GCSE (9-1)

SCIENCE

J247, J257, J248, J258, J249, J259, J250, J260

For first teaching in 2016

Exam hints for students



GCSE (9-1) Science

General Tips









Use capital letters where appropriate when writing chemical symbols. The first letter of an element symbol is always a capital.

Hydrocarbons contain carbon and hydrogen

Hydrocarbons contain carbon and hydrogen only

Use precise terminology, so your answer shows the whole picture.

CON

Longer answers don't always lead to more marks. If correct responses are contradicted, marks can be lost.

a. ~~~~ X b. ~~~~~~~~~

For MCQs, if you don't know the answer try eliminating options by annotating. Don't leave MCQ answers blank!

The value of A is greater than that of B

When a question asks you to make a comparison, make sure you clearly describe differences and/or similarities.



Concise responses are the best responses. All marks can be obtained within the answer space provided.

Your answer:



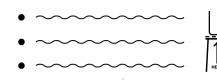
If changing the answer for an MCQ, completely cross out the wrong letter and write the correct one anew. Use upper-case letters only

Use the information in the table to describe and compare the motion of the students.

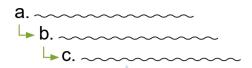
For Level of Response, answer each part of the question roughly equally. Check you have answered the whole question.

Describe and compare the bonding of the materials and suggest which of them would be best to use, giving reasons for your answer.

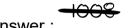
Underlining or circling key information in questions will help you remember, as will jotting down ideas and equations.



Write like a scientist, not like a storyteller. Using bullet points or diagrams can reduce the amount you have to write.



The different parts of extended questions are linked. Information and answers from part (a)ii may help with part (b)i.



Answer :

Cross out answers if you need to change them. Trying to correct an answer by writing over it can make it unclear.

Practical skills

This shows the results are reliable

This shows the results are repeatable

This shows the results are reproducible

Use of the term 'reliability' is not

encouraged. 'Repeatability', 'confidence'

and 'reproducibility' are more appropriate.

Scientific diagrams of equipment should

be schematic and factual (not three-

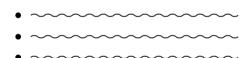
dimensional and artistic).







Remember that precision is the closeness of agreement between different results. It is



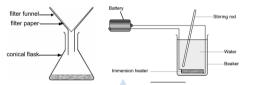
not the same as accuracy.

If describing a practical method use bullet points to give a list of simple, clear instructions someone else can follow.





Accuracy is a measure of how close a result is to the true value.



Practice applying what you know to new situations. Unfamiliar experiments will still use apparatus and techniques you know.

What would make the results more accurate?

Doing more repeats

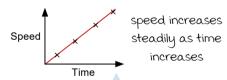


Repeats only improve precision of the set of measurements and not their accuracy.



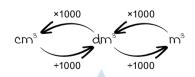


Be specific with suggested safety precautions and why they are needed.

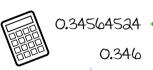


When describing data (graphs/tables) comment on trends, patterns and correlations, not just single data points.

Maths skills



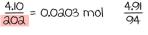
In calculations always check the units and make conversions if needed.

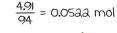


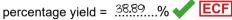
It's always more accurate to round once, for the final answer, and work with unrounded values on the calculator.



Make sure you give answers to the number of significant figures asked for after performing calculations.

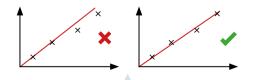




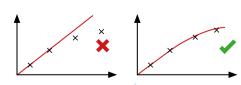




Show clear working for calculations. Error carried forward may mean a response still gains marks if a mistake is made.

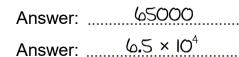


Lines of best fit should cover all points and have a fair distribution of points above and below the line.



Lines of best fit can be straight or curved. They don't have to extend to the axes or origin if not appropriate.

2



You need to be able to convert results between decimal form and standard form (e.g. $a \times 10^{n}$).

GCSE (9-1) Science

Biology



Antibodies made by our immune system recognise and bind to the antigens found on the outside of foreign organisms.



biotic abiotic

Remember living things are biotic, non-living things are abiotic.



Food chains show the direction energy moves between organisms. Pyramids of biomass show total energy in each level.



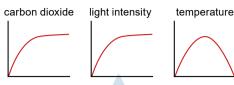




Students didn't know names of cell types and structure of organs. Practise matching diagrams of cells/organs to their functions.



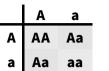
Make sure that your capital letters in a Punnett square are much bigger than the lower-case letters.



Carry out different experiments and analyse graphs to understand how limiting factors affect photosynthesis.

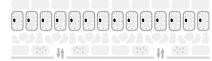


The substrate acts as a key and the enzyme as a lock. The active site is the specific part of the lock the key fits into.



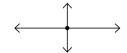


Practise Punnett squares and calculating probability of genetic diseases. Ensure you know different inheritance terms.



Many students did not understand the function of guard cells and the stomata.

Physics



A free body diagram is the scientific way to show the forces acting on an object. Most students showed artistic sketches instead.



When drawing magnetic field lines, the arrows go from the north pole to the south pole and should not overlap.



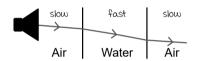
In circuit diagrams components must be connected and in the correct position. Ammeter in series, voltmeter in parallel.







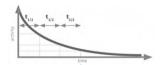
Energy can be stored or transferred between stores. There are not different types of energy - only different stores.



Sound waves move faster in denser mediums. e.g. Sound waves move faster in water than in air.



The National Grid uses step-up and stepdown transformers to reduce the current and increase voltage in transmission lines.



The half-life of a radioactive source is the time taken for half its nuclei to decay and can be found using an activity-time graph.

- 6 protons, 6 neutrons

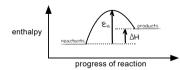
- 6 protons, 7 neutrons

Isotopes of an element have the same number of protons in the nucleus but different numbers of neutrons.

Force is equal to mass times the acceleration, so...

Using equations to help communicate your answer can be quicker than several sentences saying the same thing.

Chemistry

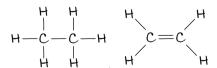


Energy profile diagram arrows are single headed, show direction of energy change and extend to the limits of the change.

Be clear as to whether an attraction is between molecules or between the atoms within a molecule.

$$mg + O_a \rightarrow mgO \times$$
 $a mg + O_a \rightarrow a mgO \checkmark$

Check equations for balancing errors after writing them. Remember that any charges should also be balanced.



Make sure you know the names of the different organic homologous series.

$$aH^+ + ae^- \longrightarrow H_a$$

 $aCl^- \longrightarrow Cl_a + ae^-$

Half equations show you what happens to each ion in the reaction showing the electrons involved.







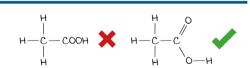
When writing the chemical formula of an ionic compound, remember the charges have to balance in ionic formulas.



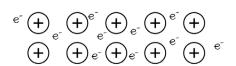
Atomic number is the smaller number: the number of protons in an atom. Atomic Mass is the larger number: the mass of an atom.



Many students could not remember the chemical tests for ions in solution.



When drawing display formulae show all the bonds in the compound.



When drawing the structure of a metal, draw the delocalised electrons surrounding and in between the metal ions.

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